TEST NAME: PSc 1.2 Density Spring 2018 TEST ID: 2181335 GRADE: 09 - Ninth Grade - 12 - Twelfth Grade SUBJECT: Life and Physical Sciences TEST CATEGORY: Shared Classroom Assessments

02/09/18, PSc 1.2 Density Spring 2018
Student:
Class:
Date:

1. An Arctic researcher determines the volume of an iceberg sample to be 100 mL and the mass to be $\mathbf{9 6} \mathbf{g}$. What is the density of the sample?
A. $\quad 104 \mathrm{~g} / \mathrm{mL}$
B. $96 \mathrm{~g} / \mathrm{mL}$
C. $1.04 \mathrm{~g} / \mathrm{mL}$
D. $\quad 0.96 \mathrm{~g} / \mathrm{mL}$
2. 



A Physical Science student is performing a science experiment.
Which conclusion can be made about the density of the liquids?
A Oil has the lowest density and dishwashing soap has the highest density.
B. Dishwashing soap has the lowest density and water has the highest density.
C. Dishwashing soap has the lowest density and oil has the highest.
D. Water has the lowest density and oil has the highest density.
3. What is the density of 180 g of methanol whose volume is 600 mL ?

A $\quad 0.3 \mathrm{~g} / \mathrm{mL}$
B. $\quad 3.33 \mathrm{~g} / \mathrm{mL}$
C. $108,000 \mathrm{~g} / \mathrm{mL}$
D. $108 \mathrm{Kg} / \mathrm{L}$
4. What is the volume of 244 grams of aluminium that has a density of $2.71 \mathrm{~g} / \mathrm{mL}$ ?
A. $\quad 90 \mathrm{~mL}$
B. 90 g
C. $\quad 661.24 \mathrm{~mL}$
D. $\quad 0.011 \mathrm{~mL}$
5. What is the mass of a substance that has a volume 3.06 mL , and the density $17.7 \mathrm{~g} / \mathrm{mL}$ ?
A. $\quad 54.2 \mathrm{~g}$
B. $\quad 54.2 \mathrm{~mL}$
C. $\quad 4.91 \mathrm{~g}$
D. $\quad 0.17 \mathrm{~g}$

